
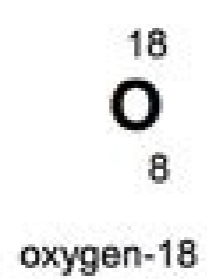
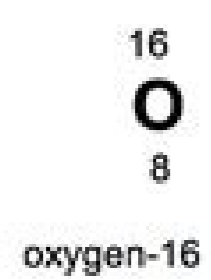


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Open



12
C
carbon
6

the relative atomic mass is always the larger of the two numbers by an element

an element in the periodic table

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Working out relative formula mass

Compound	Workings	M _r
LiF		
C ₂ H ₄		
NaHCO ₃		
Al ₂ O ₃		
H ₂ SO ₄		
CaCl ₂		
NH ₄ NO ₃		
Mo(CO) ₆		
B(CH ₂ CH ₃) ₃		

